Name:	Subject: Chemistry	Class: 12 th	Time: 60 minutes	Total Marks:	30
Chapter No.7	MJDexpert.com		Obtained marks		

Note: Please attempt any 7 short questions from Question 2. Also, attempt both parts (a and b) of Question 3. Cutting and removal of any content is strictly prohibited.

Objective-Section

Q. 1 Encircle the c	orrect answer by filling the	appropriate circle.	(8x1=8)			
The chemist who synt	hesized urea from ammonium (cyanate was				
(A) Berzelius	(B) Kolbe	(C) Wohler	(D) Lavoisier			
2. Linear shape is associated with which set of Hybrid orbital						
(A) SP	(B) SP2	(C) SP3	(D) dSP2			
Which of the following	g is an amide?					
(A) $(NH_2)_2C0$	(B) NH ₂ CH ₃	(C) $C_6H_5NH_2$	D) $N(CH_3)_3$			
4. Which set of hybrid orbital has planner triangular shape						
(A) Sp3	(B) Sp	(C) Sp2	(D) dSp2			
One of the following N	Molecule is sp2 hybridized:					
(A) CH ₃ - CH ₃	(B) $CH_2 = CH_2$	(C) CH= CH	(D) CH ₄			
6. The crude petroleum is separated in fractions by:						
(a) Filtration	(b) Fractional distillation	(c) steam distillation	(d)			
Fractions sublimation						
Carbon atom in dimet	thyl ether is					
(a) Sp-hybridized	(b) Sp ² hybridized	(c) Sp ³ -hybridized	(d) None			
8. Number of isomers of C ₄ H ₁₀ is						
(a)1	(b) 2	(c) 3	(d) 4			
	The chemist who synt (A) Berzelius Linear shape is associated (A) SP Which of the following (A) (NH ₂) ₂ CO Which set of hybrid of (A) Sp3 One of the following Management of the crude petroleum in the crude petroleum in the crude of the crude petroleum in the crude of	The chemist who synthesized urea from ammonium (A) Berzelius (B) Kolbe Linear shape is associated with which set of Hybrid (A) SP (B) SP2 Which of the following is an amide? (A) (NH ₂) ₂ CO (B) NH ₂ CH ₃ Which set of hybrid orbital has planner triangular shape (A) Sp3 (B) Sp One of the following Molecule is sp2 hybridized: (A) CH ₃ - CH ₃ (B) CH ₂ = CH ₂ The crude petroleum is separated in fractions by: (a) Filtration (b) Fractional distillation Fractions sublimation Carbon atom in dimethyl ether is (a) Sp-hybridized (b) Sp ² hybridized Number of isomers of C ₄ H ₁₀ is	Linear shape is associated with which set of Hybrid orbital (A) SP (B) SP2 (C) SP3 Which of the following is an amide? (A) $(NH_2)_2CO$ (B) NH_2CH_3 (C) $C_6H_5NH_2$ Which set of hybrid orbital has planner triangular shape (A) Sp3 (B) Sp (C) Sp2 One of the following Molecule is sp2 hybridized: (A) CH_3 - CH_3 (B) CH_2 = CH_2 (C) CH = CH The crude petroleum is separated in fractions by: (a) Filtration (b) Fractional distillation Fractions sublimation Carbon atom in dimethyl ether is (a) Sp-hybridized (b) Sp² hybridized Number of isomers of C_4H_{10} is			

Subjective-Section

Q.2 Write short answers of any SEVEN of the following questions: (7x2=14)

- 1. What are homocyclic and heterocyclic compounds, give one example of each
- 2. What are fossil fuels? How are they formed?
- 3. Define the term octane number. How octane number of gasoline is improved?
- 4. What is Atomic Orbital Hybridization?
- 5. What type of structural isomerism is shown by ethers?
- 6. What is Tautomerism? Give an example.
- 7. What is Cis-trans Isomerism? explain with the help of an example.
- 8. Why is there no free rotation around a double bond?

Q.No.3 Long Question:

(4+4=8)

- a. Define the following with their uses (i) Thermal cracking (ii) Catalytic cracking (iii) Steam cracking
- b. What is meant by orbital hybridization? Give the structures of ethane and ethyne