

Name: _____						
<b>Subject:</b> Chemistry		<b>Class:</b> 12 <sup>th</sup>		<b>Time:</b> 60 minutes		
<b>Chapter No.2</b>		<b>MJDexpert.com</b>			<b>Total Marks:</b>	<b>30</b>
				Obtained marks		

**Note:** Please attempt any 8 short questions from Question 2. Also, attempt both parts (a and b) of Question 3. Cutting and removal of any content is strictly prohibited.

### Objective-Section

**Q. 1 Encircle the correct answer by filling the appropriate circle. (8x1=8)**

- Beryllium reacts with sodium hydroxide to form**  
 (A)  $\text{BeO} + \text{NaH}$       (B)  $\text{Na}_2\text{BeO}_2 + \text{H}_2$       (C)  $\text{BeO}_2 + \text{H}_2$       (D)  $\text{Be}(\text{OH})_2 + \text{Na}$
- The oxides of beryllium are**  
 (A) Acidic      (B) Basic      (C) Ionic      (D) Amphoteric
- Which of the following does not belong to alkaline earth metals?**  
 (A) Be      (B) Ra      (C) Ba      (D) Rn
- Which of the following do not form oxide of Nitrogen on heating?**  
 (A)  $\text{LiNO}_3$       (B)  $\text{NaNO}_3$       (C)  $\text{Ca}(\text{NO}_3)$       (D)  $\text{Mg}(\text{NO}_3)_2$
- Which gas is evolved at cathode during the electrolysis of brine in diaphragm cell?**  
 (A)  $\text{H}_2$       (B) Na      (C)  $\text{Cl}_2$       (D)  $\text{O}_2$
- Natron has the chemical formula**  
 (A)  $\text{NaNO}_3$       (B)  $\text{KNO}_3$       (C)  $\text{Na}_2\text{CO}_3 \cdot \text{H}_2\text{O}$       (D)  $\text{CaCO}_3$
- The element deposited at Cathode during electrolysis of brine in Diaphragm Cell is:**  
 (A)  $\text{Cl}_2$       (B) Na      (C)  $\text{O}_2$       (D)  $\text{H}_2$
- In Nelson's cell for the production of NaOH cathode is made up of**  
 (a) Mercury      (b) Graphite      (e) Steel vessel lined inside with asbestos      (d) Iron

### Subjective-Section

**Q.2 Write short answers of any eight (8) of the following questions: (8x2=16)**

- Write formulas of (i) Colemanite (ii) Cryolite
- What is the milk of magnesia? Where it is used?
- What happens when we heat  $\text{Li}_2\text{CO}_3$  and  $\text{CaCO}_3$  separately?
- Discuss in which respect  $\text{KO}_2$  can be used by mountaineers.
- Why the aqueous solution of  $\text{Na}_2\text{CO}_3$  is alkaline?
- Give two advantages of Down's cell process.
- How lime Mortar is prepared?
- What is the chemical nature of lime water and milk of magnesia?
- Why lime is added to acidic soil?

**Q.No.3 Long Question: (4+4=8)**

- Write a note on the peculiar behavior of Beryllium
- Describe the two problems involved in the manufacture of caustic soda by Nelson cell and how these problems were solved?